Y8 Science Term 6		Variation, Electricity and Atomic Structure					
Week 1: Term 5 Recap		Week 2: DNA		Week 3: Variati	on Trinity Leeds		
energy source	a source from which useful energy can be extracted or converted	DNA	Deoxyribonucleic acid. The material inside the nucleus of cells in a double	variation	The differences between individuals of the same species		
non-renewable	an energy source that will not run out an energy source that is used faster than it is replenished and will run out the amount of energy transferred in a	chromosome	stranded helix structure. The structure made of DNA that codes for all the characteristics of an organism A section of DNA which codes for	heredity	The passing on of characteristics genetically from one generation to another		
set amount of time Renewable and Non-Renewable Energy Sources Renewable energy Solar Non-renewable energy Solar		nucleotide	proteins The monomer of DNA made up of sugar, phosphate and one of four bases (A,T,C or G)	genetic variation environmental variation ightarrow ightarrow igh	Variation caused by genes e.g. Eye colour or ear lobe shape Variation caused by the environment (e.g. culture/ weather). Examples include skin colour, religion, tattoos		
Hydropower Geothermal Wind Nuclear Natural gas		Deoxyribose		fertilisation	When a female gamete (sex cell) and male gamete (sex cell) nuclei fuse to produce a zygote.		
joules (J) kWh	the units for all types of energy the unit used to state the amount of energy used by a 1kW appliance for 1 hour	 There are human be The four pairs with Bases are 	23 pairs of chromosomes in each dy cell (46 in total). NA bases are A,T,C and G – A always T and C always pairs with G held together by hydrogen bonds.	Ferti	tilization		
compare	to find similarities and differences between to objects			46 Chromosomes in 23 pairs Sperm 23 Chromosomes			
	Extension QR Codes - R	ead the BBC bitesi	ize new knowledge page, watch the video,	and complete the s	self-quiz		

Y8 Science Term 6		Variation, Electricity and Atomic Structure					Trinity			
Week 4: Circuits		Week 5: Resistance		Week 6: Atomic Structure			Leeds			
circuit series circuit \bigcirc \bigcirc \bigcirc parallel circuit \bigcirc \bigcirc current ammeter $-\bigcirc$ \frown	A complete loop which allows an electric current flow. A circuit with one loop through which current flows. A circuit with more than one loop which current can flow. The rate of flow of charge, measured in Amps (A) A device, connected in series, which measures the current in a circuit.	resistance ohms (Ω) variable resistance equation	How much the wires and components reduce the flow of charge The unit of measure for resistance A factor which could affect experimental results (can be independent, dependent or control), Resistance = Potential difference (V) ÷ current(A)	Electron Proton Neutron atom mass number atomic number ion	The smallest pie up of protons, e neutrons. The total number neutrons in the The number of p nucleus of an at An atom that has electrons.	ce of matter, lectrons, and er of protons a nucleus of an protons only i om. as lost or gaine	e of matter, made ctrons, and of protons and ucleus of an atom. otons only in the n. lost or gained			
voltmeter	A measure of the difference in energy between two parts of a circuit measured in volts (V). A device, connected parallel to a component, which measures potential difference.	Increasing the re make it harder fo		Subatomic particle proton neutron electron	Relative mass11Negligible	Relative charge1+01-				
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